

Effect of Alloy Composition on the Oxidation Behaviour and Cr Vaporization of High-Cr Steels for SOFC Cathode Air Pre-heater

*Kun Zhang,
Robert Steinberger-Wilckens*

Centre for Fuel Cell & Hydrogen Research
School of Chemical Engineering
University of Birmingham



Introduction Cathode Air Pre-Heater (CAPH)

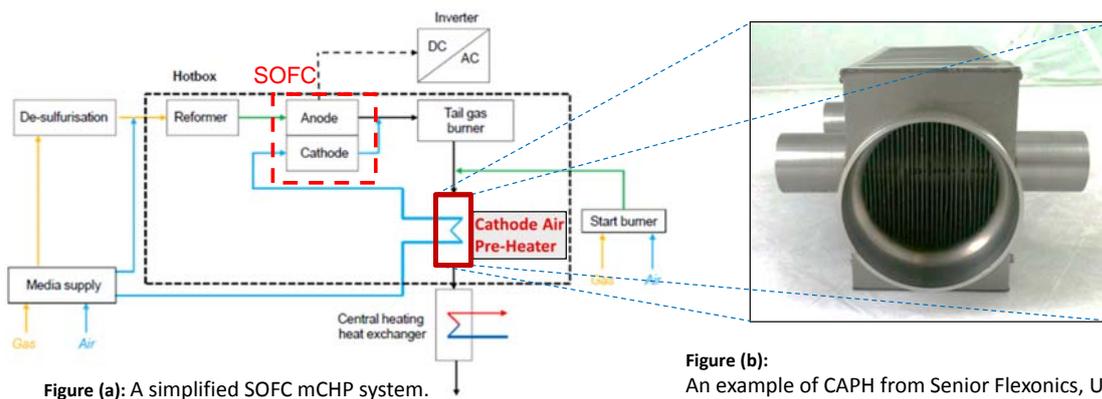
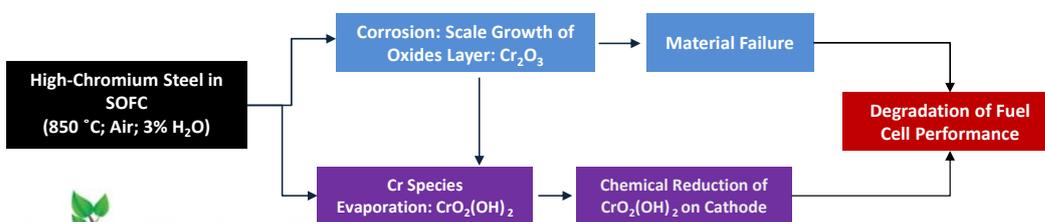


Figure (a): A simplified SOFC mCHP system.

Figure (b):
An example of CAPH from Senior Flexionics, UK.

Objectives:

- Cost reduction targets will be focused on changing from Inconel to AluChrom;
- Measure the chromium leakage (and high temperature corrosion) of Inconel 625 and AluChrom 318 supplied by VDM Metals;
- Understand the effects of surface aluminisation on the Cr retention capability of SS309.



Material: Inconel 625; AluChrom 318; Uncoated SS309; Aluminised SS309 (1 μm Al).

(wt.%)	Fe	Cr	Mn	Al	Ni	Si	Nb	W	Co	others
Inconel 625	5.0	20-23	0.5	0.4	Bal.	0.5	4.15	-	1.0	Ti 0.4; Mo 8-10; P 0.015; S 0.015
AluChrom 318	Bal.	18.8	0.21	3.58	0.24	0.32	0.73	2.02	--	Hf 0.06; Y 0.07; Zr 0.03; Cu 0.03 C 0.01; N 0.01
SS309	Bal.	22-24	2.0	--	12-15	0.75	--	--	--	C 0.2; P 0.045; S 0.03
Aluminised SS309	1 μm aluminium coated was deposited on SS309 surface by PVD technique.									

• **Experiment:**

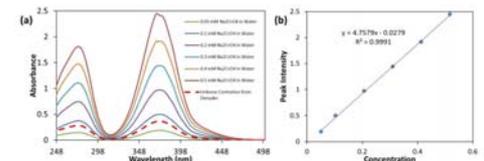
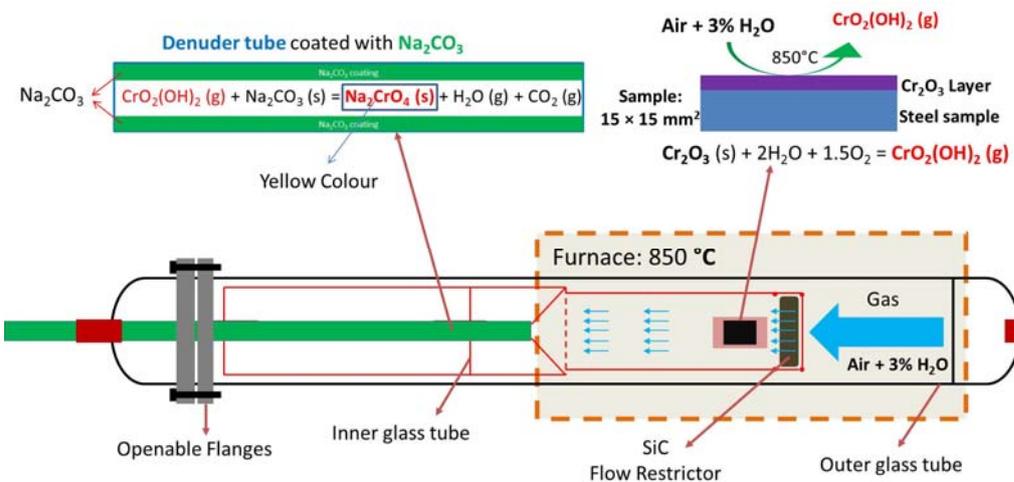
- High Temperature Oxidation Test (long-term exposure): **Normal Tubular Glassware**
- Quantification of Cr Evaporation (short-term Cr quantification): **Denuder Technique**

• **Test Condition:**

- 850 °C; 6.0 L/min Air Flow; 3 vol% H₂O

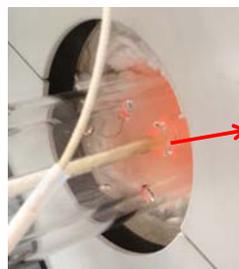


Denuder Technique for Cr Quantification



- Na₂CO₃ coating in denuder tube is washed off with DI water and quantified using UV-vis spectrophotometer;
- pure Cr₂O₃ pellets are used as Cr volatile source to confirm accuracy of denuder setup under SOFC conditions;
- efficiency of the denuder setup for Cr volatiles collection is confirmed to be around 98%.

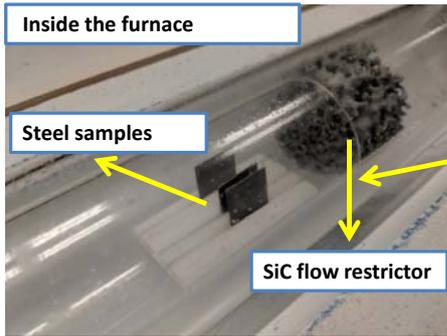
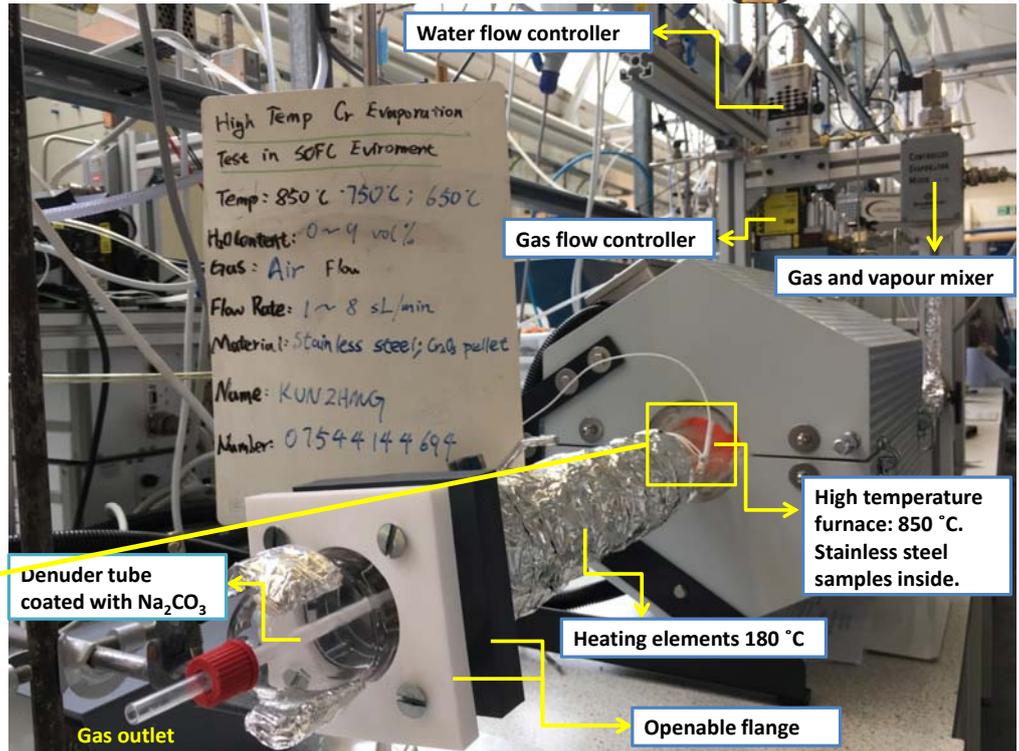
Reference:
J. Froitzheim, H. Ravash, E. Larsson, L.G. Johansson, J.E. Svensson, *J Electrochemical Society* 157 (2010) B1295.



When Cr species is collected, the denuder tube will turn yellow.



Denuder Technique for Cr Quantification – Laboratory Setup



High Temperature Oxidation Test

Test Condition:

Temp: 850 °C;
 Gas: Air + 3% H₂O;
 Flow rate: 6.0 L/min;
 Time: 2,000 hours

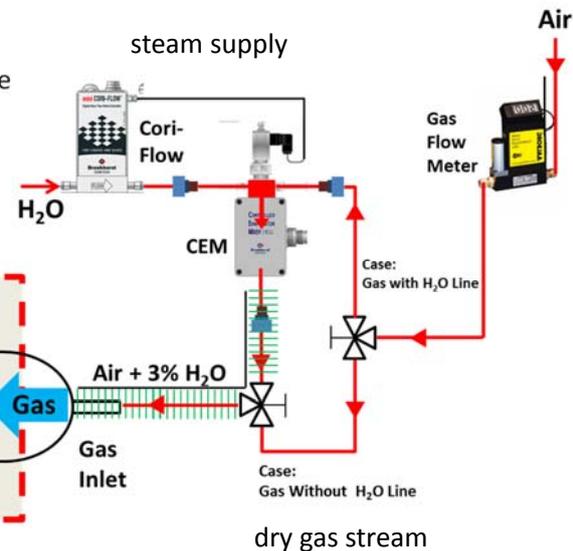
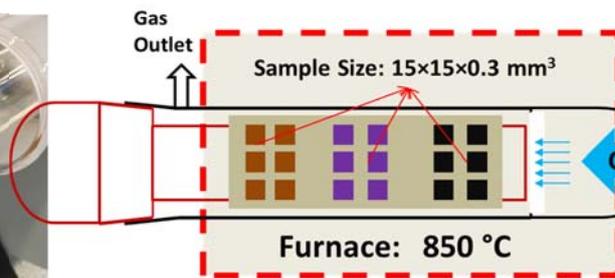
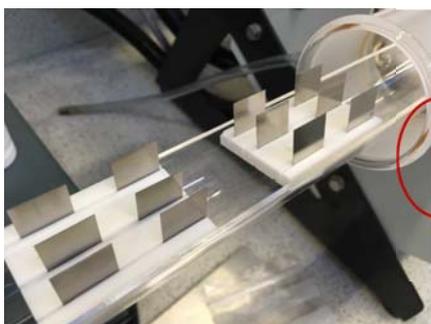
Materials:

- Inconel 625;
- AluChrom 318
- SS309
- Aluminised SS309

Post-analysis:

weighing; surface SEM/EDX;
 cross-section SEM/EDX; surface XRD.

- For HT oxidation tests, all different kinds of samples are exposed in one furnace (as shown below).
- HT oxidation test is not continuous.
- The samples in the furnace will be taken out regularly for post-test analysis.



Mass Measurement & Cr Evaporation

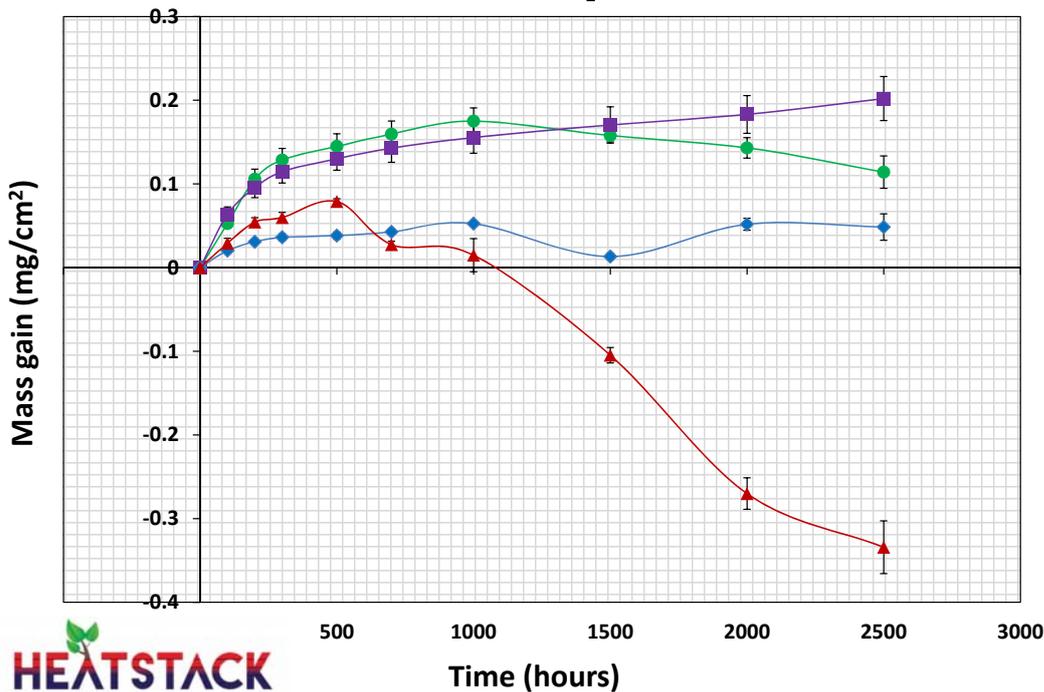
Material: Inconel 625; AluChrom 318; Uncoated SS309; Aluminised SS309.

Test Conditions: 850 °C; 6.0 L/min Airflow; 3 vol% H₂O.

Equipment: Normal tubular glassware for high temperature corrosion test;
Denuder Technique for evaporated Cr collection.

Mass Gain Measurements (High temperature oxidation test)

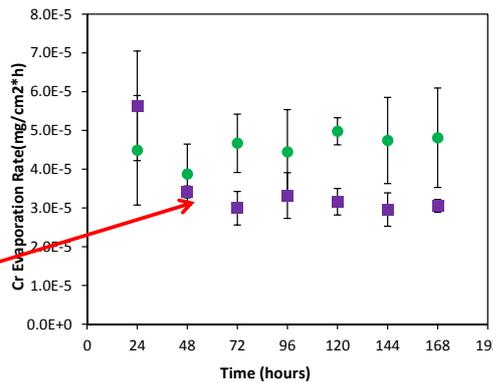
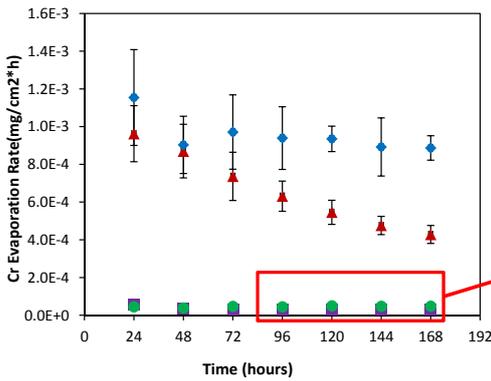
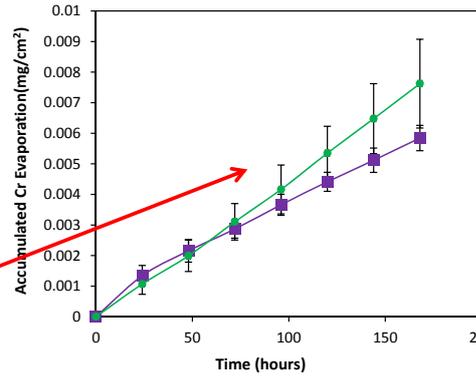
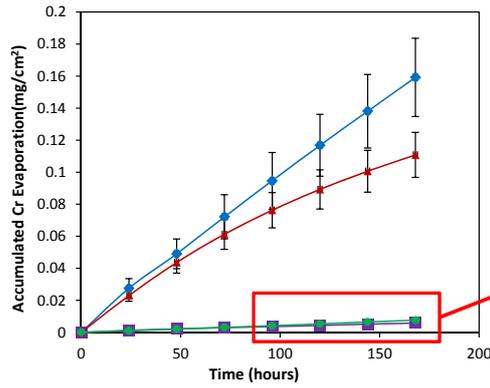
Discontinuous Gravimetric Measurements
6.0 L/min, 3 vol% H₂O, 850 °C



1. mass gain of AluChrom 318 increases even beyond 2,500 hours, which means aluminium oxide keeps growing on the surface.
2. mass gain of aluminised 309 decreases after 1,000 hours, which means the aluminium oxide spalls from the surface.
3. Inconel 625 has a relatively lower mass gain indicating that the Cr evaporation is very high.



Continuous Cr Evaporation Measurements



- 3 repeats for all samples;
- Error obtained for every sample;
- Results from Denuder testing are reliable and repeatable;
- Aluminisation for SS309 could effectively prevent Cr evaporation, but not in the long term (with the method used here).
- The results shows that AluChrom 318 has the lowest amount of Cr evaporation among all these materials.



Microstructure Analysis - SEM & EDX

Material: Inconel 625; AluChrom 318; Uncoated SS309; Aluminised SS309.

Exposure time: 0 hours; 100 hours; 500 hours; 1000 hours.

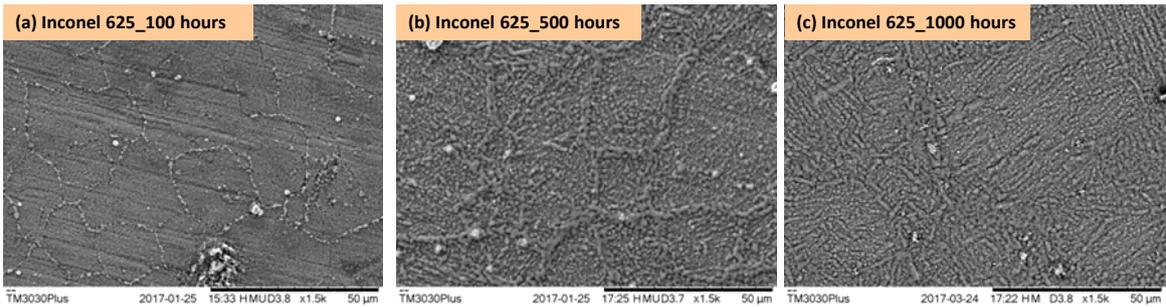
Include: Surface SEM images; Cross section SEM

Surface elemental concentration;

EDX line scan; EDX mapping;

X-ray diffraction

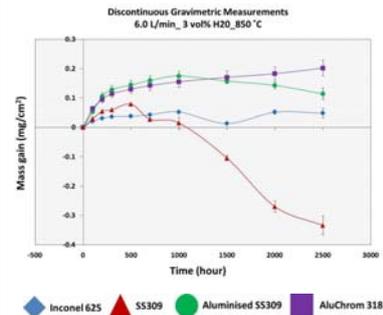
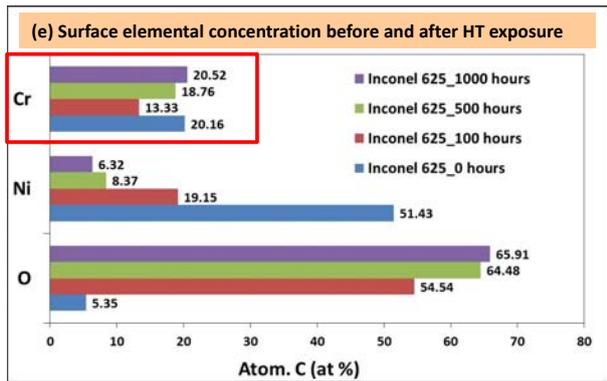
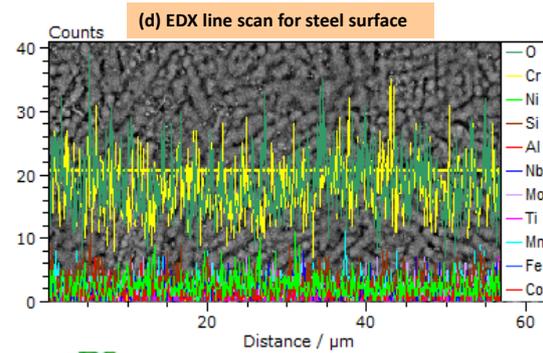
SEM/EDX: Inconel 625



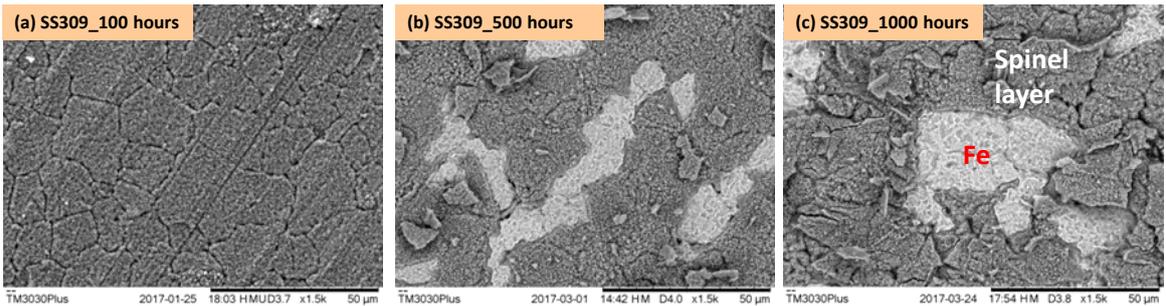
The EDX line scan confirms that the surface of Inconel 625 is completely covered by Cr_2O_3 .

With the increase of exposure time at high temperature, more Cr_2O_3 scale grows on the steel surface.

The concentration of Cr on the steel surface also increases with the increase of exposure time, which is consistent with the increase of O.



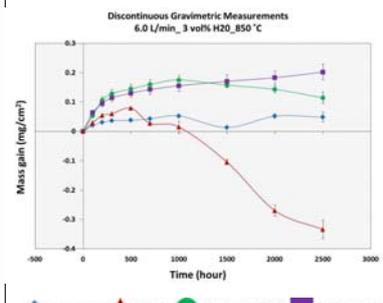
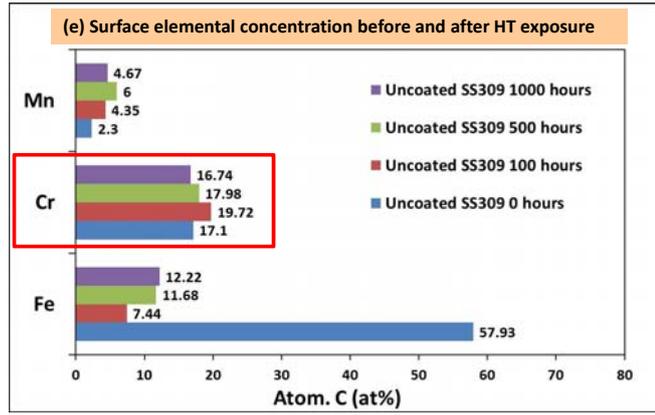
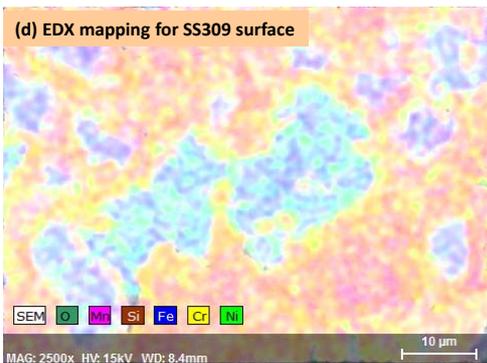
SEM/EDX: SS309



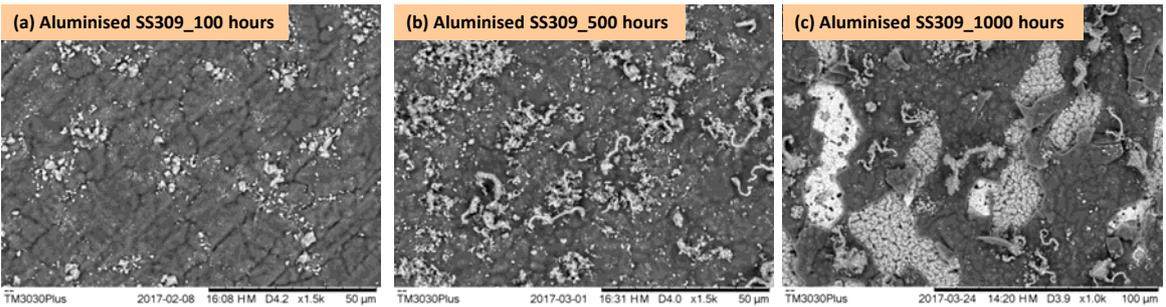
The spinel layer is formed on the surface after 100 hours of exposure at high temperature.

However, this material suffered serious spallation of the oxide layer after 500 hours exposure. Fe started to be exposed on the surface.

In figure (e), the concentration of Cr decreases with the increase of exposure time, because this material is losing Cr due to the spallation of the spinel layer.



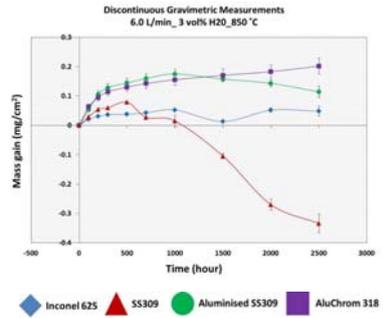
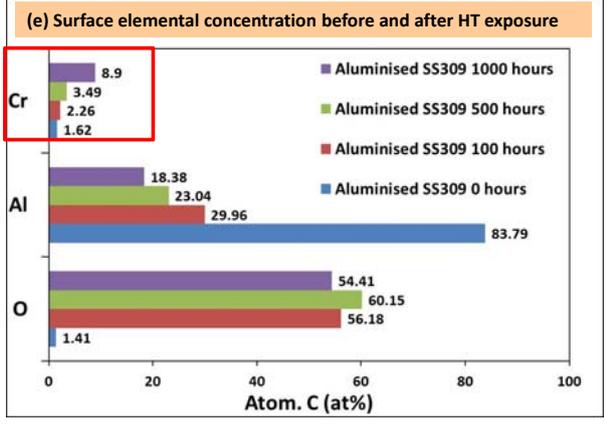
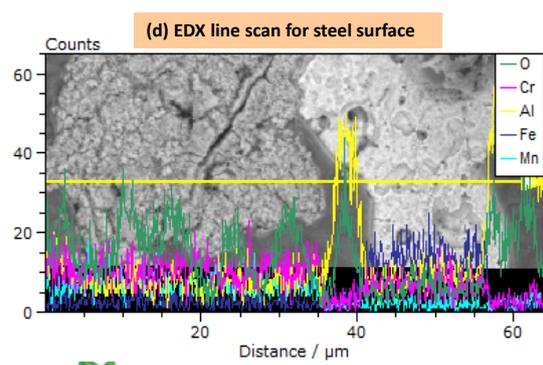
SEM/EDX: Aluminised SS309



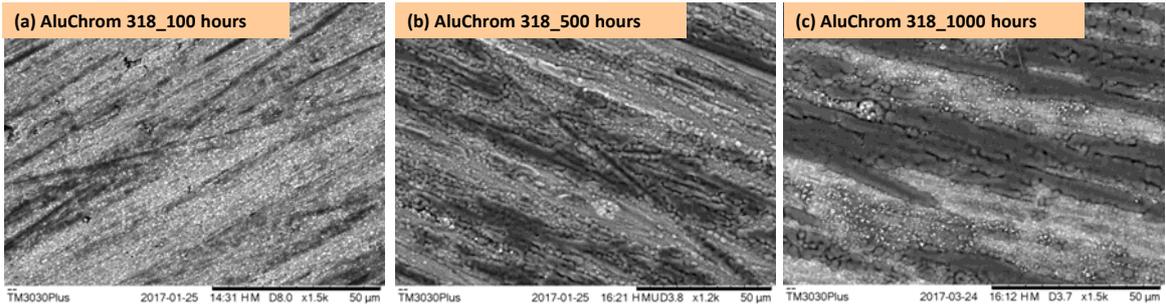
The Al formed Al_2O_3 on the surface after 100 hours of HT exposure.

The Al_2O_3 coating starts to crack after 1000 hours exposure.

In figure (e), the surface concentration of Cr subsequently increases to 8.9%.



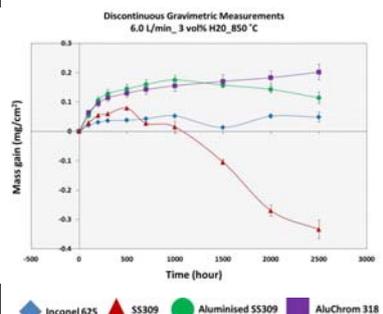
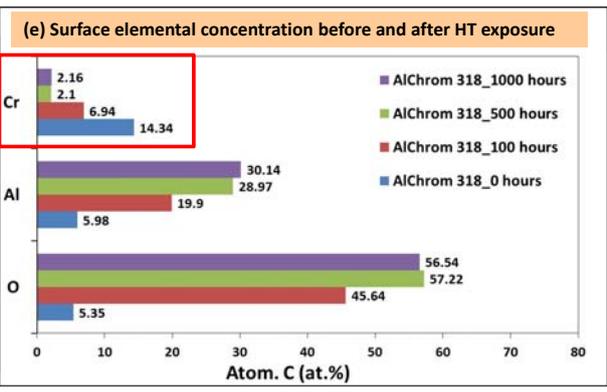
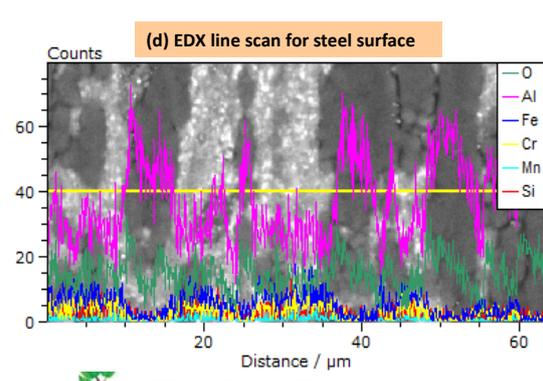
SEM/EDX: AluChrom 318



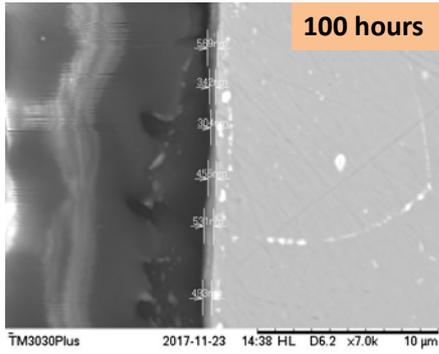
The EDX line scan confirms the steel surface is covered with Al_2O_3 .

The thickness of the Al_2O_3 layer increases with the length of exposure at 850 °C.

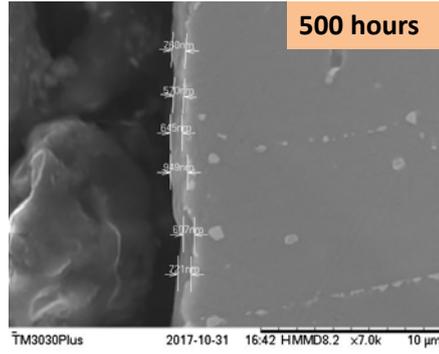
The concentration of Cr on the steel surface remains at 2.16% after 1000 hours exposure at 850 °C.



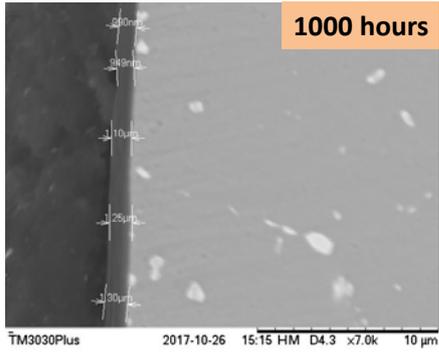
Cross-section SEM of AluChrom 318



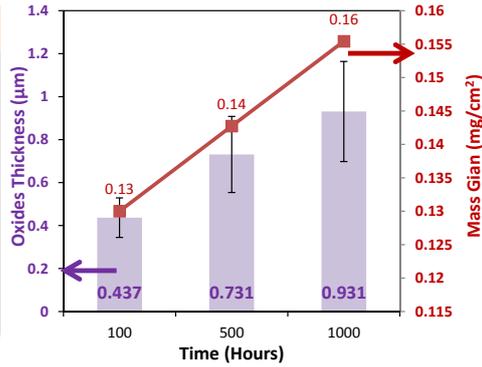
100 hours



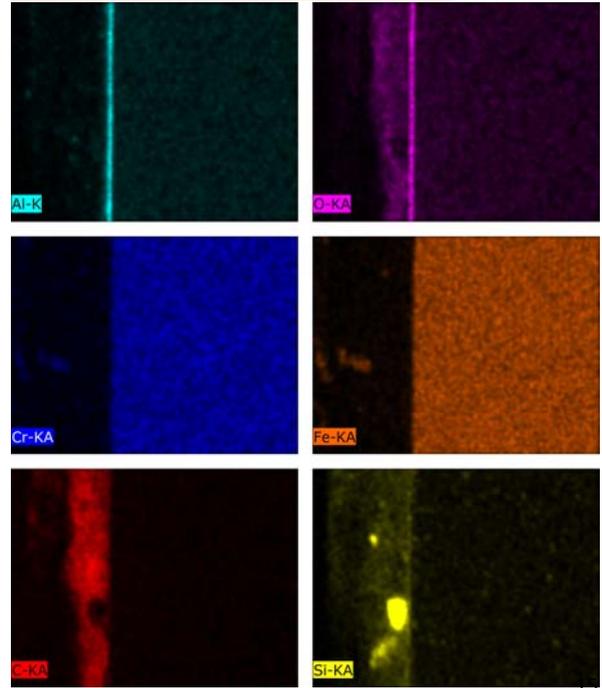
500 hours



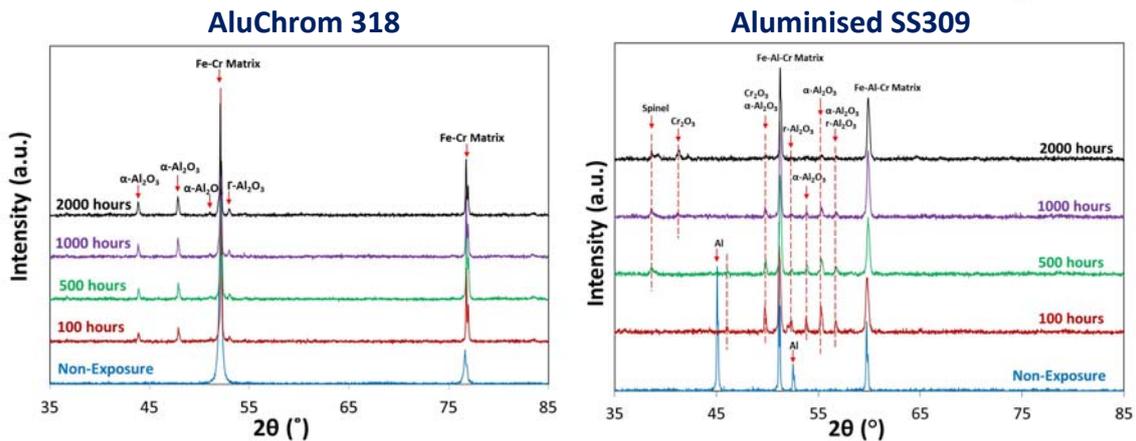
1000 hours



EDX mapping 1000 hours sample



XRD Analysis for Al-containing Steels

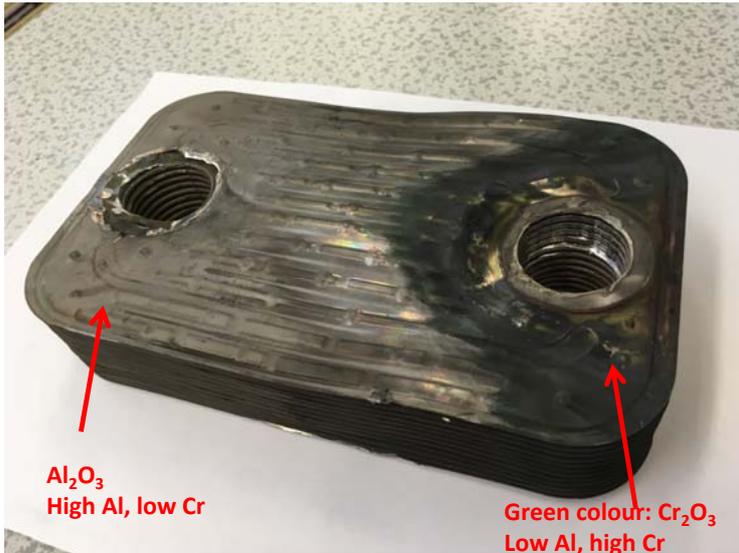


- The formation of a surface scale of $\alpha\text{-Al}_2\text{O}_3$ was observed for both materials.
- The peak intensity of $\alpha\text{-Al}_2\text{O}_3$ for AluChrom 318 shows an increasing trend.
- The XRD peaks of $\alpha\text{-Al}_2\text{O}_3$ for aluminised 309 almost disappeared after 2000 hours.
- For aluminised 309, the peaks spinel and Cr_2O_3 started to appear after 500 hours exposure.
- The alumina layer formed on AluChrom 318 is much stronger than that on aluminised 309.

Heat Exchanger from Vaillant

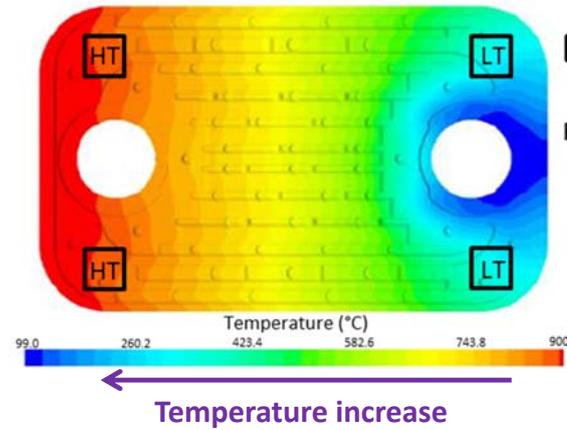


Steady state; AluChrom 318;
5800 hours; old design



Hot Zone
900 °C

Cold Zone
100 °C



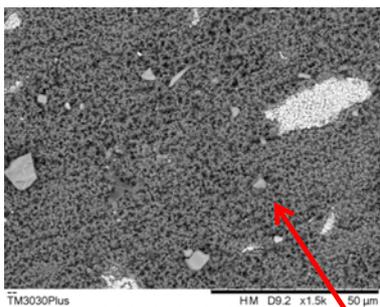
- The green colour on the left image is chromium oxide. This region is the cold zone and the temperature is about 100 °C. The temperature is not high enough for the Al₂O₃ to form, so the Cr₂O₃ forms on the surface first.
- The hot zone shows an extremely low concentration of Cr and high concentration of Al, which means the plate surface is completely covered with a very thick Al₂O₃ coating.
- A (pre-)heat treatment procedure is still required for the AluChrom 318.

Heat Exchanger from Vaillant

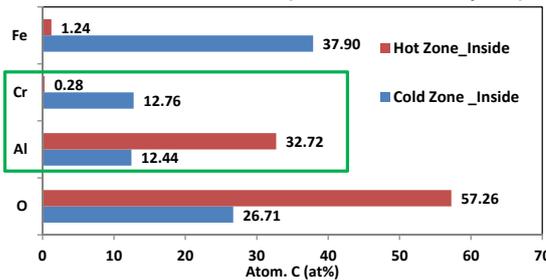


Steady state; AluChrom 318;
5,800 hours; old design

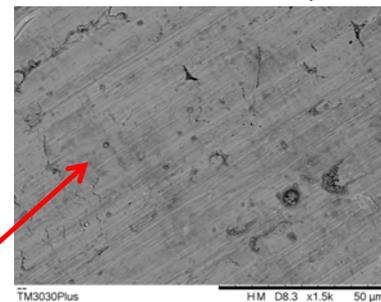
SEM for **hot** zone of inside plate



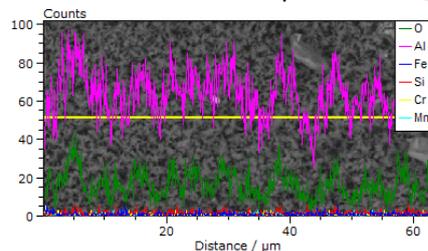
Comparison of the elemental concentration between hot zone and cold zone (inside of the stack plate)



SEM for **cold** zone of inside plate



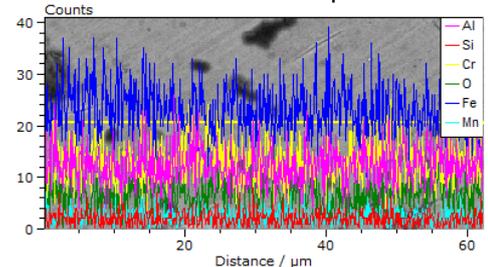
EDX line scan for **hot** zone of inside plate



Hot Zone 700 °C ← Temperature increase → Cold Zone 100 °C



EDX line scan for **cold** zone of inside plate

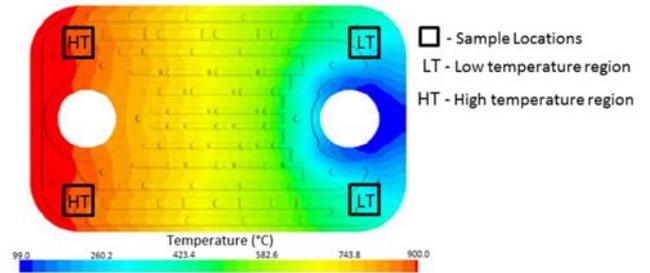
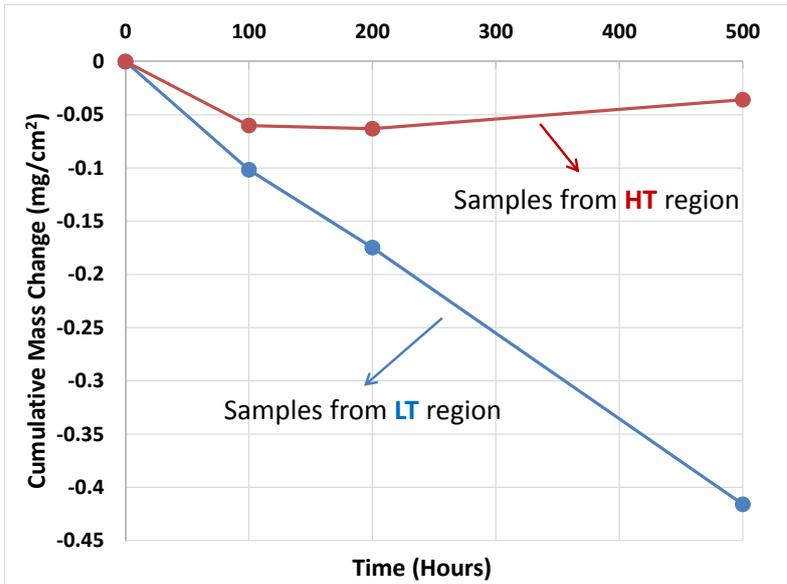


Mass Gain Measurements

850 °C, 6.0L/min air flow + 3 vol% H₂O



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Sample from HT regions:

- Thick alumina layer covered both inside and outside of the plate;
- The slight decrease of mass before 200 hours maybe due to a small amount of alumina spallation.

Sample from LT regions:

- Outside is covered with thick Cr₂O₃, inside is covered with thin alumina.
- The large amount of mass decrease is due to the large amount of Cr evaporation from outer and internal surface of plate.

Since the alumina coverage of the internal face of plate in the cold zone is not enough, it will cause much high Cr evaporation compared to the internal face of plate in the hot zone. Therefore, pre-heat treatment in air is necessary to prevent Cr₂O₃ formation in both hot and cold zone.

19

Conclusions



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- **Inconel 625** shows the **largest amount of Cr evaporation** because the surface is totally covered with Cr₂O₃.
- Uncoated **SS309** also exhibits serious **Cr leakage** due to the spallation of the spinel layer ((Mn, Cr)O_x).
- For **AluChrom 318**, the α-Al₂O₃ formed on the surface could effectively **prevents Cr evaporation** in the long term.
- For **aluminised SS309**, the α-Al₂O₃ produced by PVD can dramatically reduce Cr evaporation in the beginning. However, the spallation of the Al₂O₃ coating was detected after long term exposure, followed by **increased Cr leakage**.
- A heat treatment process is needed for the heat exchanger plate prior to operation.



Future Work

- For Cr evaporation tests:
 - To run short term (one week) Cr evaporation test on **welded AluChrom 318** (repeat 3 times),
 - To run **long term (1,000 hours)** Cr evaporation test for Inconel 625, SS309, aluminised SS309, AluChrom 318 and welded AluChrom 318.
- For high temperature oxidation tests:
 - Aim to complete 10,000 hours.
- Post- test analysis:
 - Cross-section SEM/EDX for Inconel 625, SS309 and aluminised SS309 (include: **0, 100, 500 and 1,000 h** samples),
 - Surface SEM/EDX for Inconel 625, SS309, aluminised SS309 and AluChrom 318 (including **5,000 and 10,000 h** samples).
- Heat exchanger plates from Vaillant
 - To find a temperature for **pre-heat process** of heat exchanger plate,
 - To analyse chromium evaporation and weight gain of 'used' heat exchanger plates.



Thank you for your attention!
Any Questions?

Kun Zhang

KXZ295@student.bham.ac.uk

Robert Steinberger-Wilckens

